

## NOMENCLATURE PRACTISE (DO IT!!!)

Name	Formula	Formula	Name
manganese (II) hydroxide		Na <sub>2</sub> SO <sub>3</sub>	
copper (II) silicate		Ni(OH) <sub>3</sub>	
iron (III) chloride		FeClO <sub>4</sub>	
nickel (III) sulfite		Al(NO <sub>2</sub> ) <sub>3</sub>	
calcium cyanide		Ag <sub>2</sub> SO <sub>3</sub>	
potassium thiocyanate		ZnMnO <sub>4</sub>	
calcium thiocyanate		NaClO	
sodium hypochlorite		KBrO <sub>2</sub>	
ammonium sulfate		CuPO <sub>3</sub>	
ammonium carbonate		Co(NO <sub>2</sub> ) <sub>3</sub>	
magnesium hydroxide		Fe(OH) <sub>3</sub>	
nickel (III) silicate		Ba(CrO <sub>4</sub> ) <sub>2</sub>	
lithium nitrite		K <sub>2</sub> Cr <sub>2</sub> O <sub>7</sub>	
aluminum sulfite		NaClO <sub>4</sub>	
lead (IV) chromate		Li <sub>2</sub> S <sub>2</sub> O <sub>3</sub>	
potassium perbromate		Al(OH) <sub>3</sub>	
calcium permanganate		Na <sub>2</sub> SiO <sub>3</sub>	
nickel (III) hydroxide		Mg(ClO <sub>2</sub> ) <sub>2</sub>	
copper (II) nitrite		Zn(NO <sub>2</sub> ) <sub>2</sub>	
sodium iodite		ZnSO <sub>3</sub>	
potassium hypochlorite		AlPO <sub>4</sub>	

# NAMING PRACTISE

Name	Formula	Formula	Name
lithium phosphite		$K_2SO_3$	
copper (II) hydroxide		$Ni(NO_2)_2$	
iron (II) permanganate		$FePO_3$	
nickel (II) chromate		$Al(OH)_3$	
mercury (I) dichromate		$Ag_2CrO_4$	
titanium (IV) bromite		$Cu_2CO_2$	
calcium perchlorate		$SnCr_2O_7$	
ammonium nitrate		$KBrO_2$	
barium hydroxide		$Cu(ClO_4)_2$	
tin (II) silicate		$K_2SiO_3$	
lead (IV) phosphate		$NaCN$	
nickel (III) nitrite		$Ba(OH)_2$	
lithium sulfite		$Ni_3(CO_3)_2$	
aluminum cyanide		$HClO_3$	
lead (IV) nitrate		$(NH)_4CO_3$	
potassium perchlorate		$Hg_2CO_3$	
calcium hypochlorite		$KIO_2$	
chromium (IV) iodite		$Mg(BrO_4)_2$	
copper (II) carbonite		$Zn(MnO_4)_2$	
sodium iodite		$ZnSO_3$	
potassium hydroxide		$Al_2(PO_3)_3$	